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Key indicators

Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.048  
 $wR$  factor = 0.132  
Data-to-parameter ratio = 13.6

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

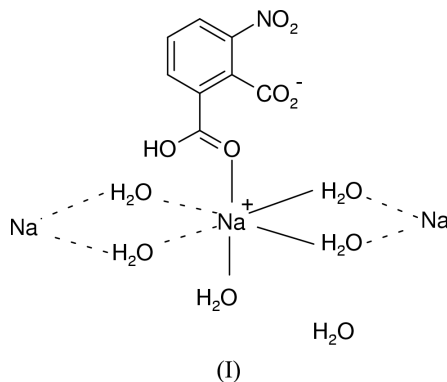
catena-Poly[[[aqua(2-carboxy-6-nitrobenzoato- $\kappa$ O)-sodium]-di- $\mu$ -aqua] monohydrate]

The title compound,  $\{[\text{Na}(\text{C}_8\text{H}_4\text{NO}_6)(\text{H}_2\text{O})_3]\cdot\text{H}_2\text{O}\}_n$ , forms polymeric chains by way of edge-sharing *via* pairs of water molecules between  $\text{NaO}(\text{H}_2\text{O})_5$  octahedra. The chains are crosslinked by  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds, resulting in a three-dimensional network.

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Comment

The crystal structure of the title sodium complex, (I), is reported here. The  $\text{Na}^+$  cation is surrounded by five water molecules and one carboxyl O atom, in an approximately octahedral geometry (Table 1). Adjacent octahedra are linked into a polymeric chain by way of edge-sharing *via* pairs of water molecule O atoms. This results in  $\text{Na1}\cdots\text{Na1}^{\text{i}}$  and  $\text{Na1}\cdots\text{Na1}^{\text{ii}}$  (see Table 1 for symmetry codes) separations of 3.4642 (19) and 3.4876 (19) Å, respectively, and an  $\text{Na1}^{\text{i}}\cdots\text{Na1}\cdots\text{Na1}^{\text{ii}}$  angle of 170.09 (7)°.



The crystal structure of (I) is stabilized by a number of intra- and interchain  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds (Table 2 and Fig. 2).

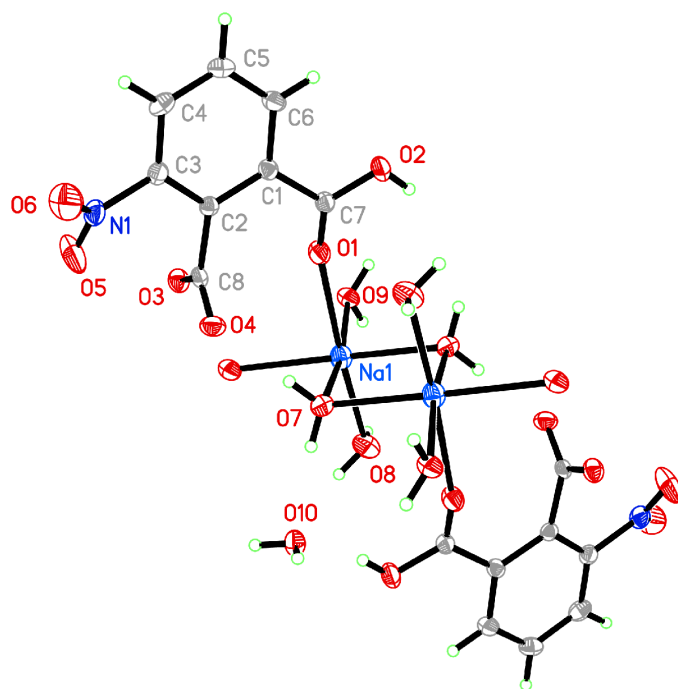
Experimental

Crystals of (I) were obtained from a 1:1 aqueous solution of 3-nitrophthalic acid and sodium hydroxide by slow concentration for 6 d.

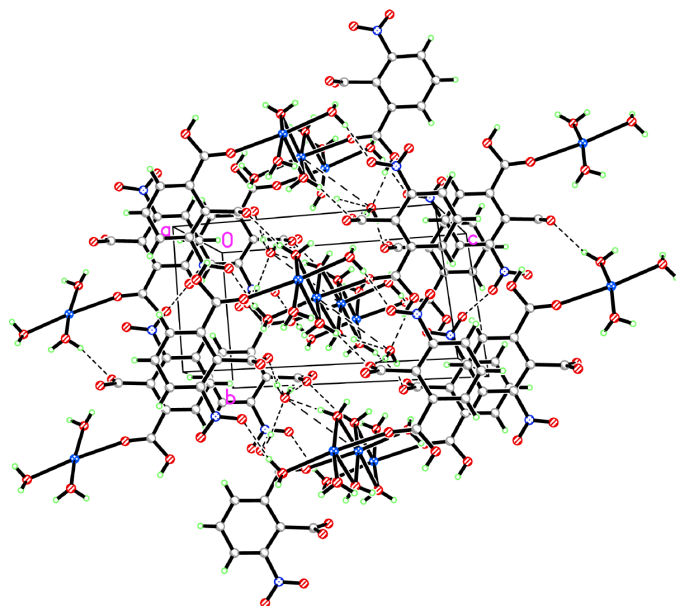
Crystal data

$[\text{Na}(\text{C}_8\text{H}_4\text{NO}_6)(\text{H}_2\text{O})_3]\cdot\text{H}_2\text{O}$   
 $M_r = 305.18$   
Triclinic,  $P\bar{1}$   
 $a = 6.9258$  (19) Å  
 $b = 7.513$  (2) Å  
 $c = 12.907$  (4) Å  
 $\alpha = 77.622$  (8)°  
 $\beta = 83.212$  (9)°  
 $\gamma = 74.103$  (8)°  
 $V = 629.6$  (3) Å<sup>3</sup>

$Z = 2$   
 $D_x = 1.610$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 766 reflections  
 $\theta = 3.0-26.2$ °  
 $\mu = 0.18$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
Block, colorless  
0.26 × 0.22 × 0.16 mm



**Figure 1**  
View of (I), showing 30% probability displacement ellipsoids. The unlabeled atoms are generated from labeled atoms by the symmetry operation  $(1 - x, 1 - y, 1 - z)$ .



**Figure 2**  
Packing diagram of (I), showing hydrogen bonds as dashed lines.

#### Data collection

Bruker SMART CCD area-detector diffractometer	2536 independent reflections
$\varphi$ and $\omega$ scans	1968 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.019$
$T_{\text{min}} = 0.916, T_{\text{max}} = 0.972$	$\theta_{\text{max}} = 26.4^\circ$
3650 measured reflections	$h = -8 \rightarrow 8$
	$k = -9 \rightarrow 8$
	$l = -16 \rightarrow 10$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.132$   
 $S = 1.11$   
 2536 reflections  
 186 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0497P)^2 + 0.4847P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.002$   
 $\Delta\rho_{\text{max}} = 0.38 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.26 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: *SHELXTL*  
 Extinction coefficient: 0.093 (8)

**Table 1**

Selected geometric parameters ( $\text{\AA}, ^\circ$ ).

Na1—O7	2.373 (2)	Na1—O7 <sup>i</sup>	2.333 (3)
Na1—O9	2.386 (2)	Na1—O9 <sup>ii</sup>	2.435 (3)
Na1—O8	2.437 (2)	O7—Na1 <sup>i</sup>	2.333 (2)
Na1—O1	2.538 (2)	O9—Na1 <sup>ii</sup>	2.435 (2)
O7 <sup>i</sup> —Na1—O7	85.18 (8)	O7—Na1—O8	102.85 (8)
O7 <sup>i</sup> —Na1—O9	98.15 (8)	O9—Na1—O8	86.90 (7)
O7—Na1—O9	169.63 (8)	O9 <sup>ii</sup> —Na1—O8	90.81 (8)
O7 <sup>i</sup> —Na1—O9 <sup>ii</sup>	173.92 (8)	O7 <sup>i</sup> —Na1—O1	85.94 (7)
O7—Na1—O9 <sup>ii</sup>	88.98 (7)	Na1 <sup>i</sup> —O7—Na1	94.82 (8)
O9—Na1—O9 <sup>ii</sup>	87.33 (7)	Na1—O9—Na1 <sup>ii</sup>	92.67 (7)
O7 <sup>i</sup> —Na1—O8	92.13 (8)		

Symmetry codes: (i)  $1 - x, 1 - y, 1 - z$ ; (ii)  $-x, 1 - y, 1 - z$ .

**Table 2**

Hydrogen-bonding geometry ( $\text{\AA}, ^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O10—H10B <sup>iii</sup> ···O4 <sup>iii</sup>	0.85	1.89	2.722 (3)	166
O10—H10A···O3 <sup>iv</sup>	0.85	1.92	2.749 (3)	165
O9—H9B···O4 <sup>v</sup>	0.85	2.11	2.939 (3)	166
O9—H9A···O3 <sup>ii</sup>	0.85	1.89	2.740 (3)	173
O8—H8B···O3 <sup>iv</sup>	0.85	2.03	2.846 (3)	160
O8—H8A···O5 <sup>iv</sup>	0.85	2.12	2.962 (3)	171
O7—H7B···O4	0.85	1.99	2.834 (3)	176
O7—H7A···O10	0.85	1.96	2.809 (3)	173
O2—H2···O10 <sup>i</sup>	0.85 (1)	1.783 (12)	2.631 (3)	169 (3)

Symmetry codes: (i)  $1 - x, 1 - y, 1 - z$ ; (ii)  $-x, 1 - y, 1 - z$ ; (iii)  $1 - x, 2 - y, 1 - z$ ; (iv)  $-x, 2 - y, 1 - z$ ; (v)  $x, y - 1, z$ .

Atom H2 was refined with an O—H distance restraint of 0.859 (10)  $\text{\AA}$ . All other H atoms attached to oxygen were found in difference Fourier maps. During refinement, they were treated as riding, with O—H = 0.85  $\text{\AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{O})$ . The H atoms attached to carbon were placed in idealized positions and refined as riding, with C—H = 0.93  $\text{\AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2001); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

#### References

- Bruker (1997). *SMART* (Version 5.051) and *SAINTE* (Version 5.A06). Bruker AXS Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (2001). *SHELXTL*. Version 6.12. Bruker AXS Inc., Madison, Wisconsin, USA.